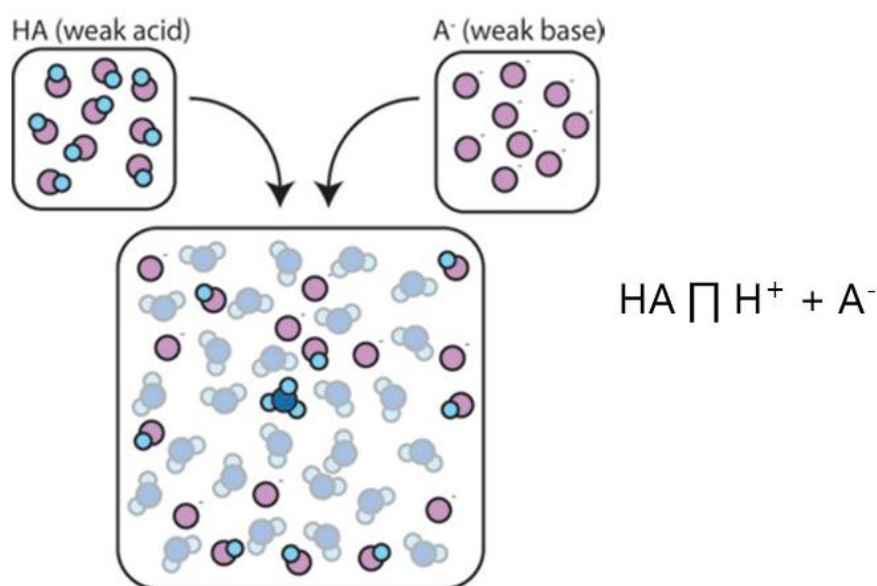


Buffer solution

Introduction: Buffer solutions resist changes in pH on the addition of a small amount of an acid or a base.

Buffer Solutions

- DEFINITION: A buffer solution contains a weak acid mixed with its conjugate base (or weak base and conjugate acid)
- Buffers resist changes in pH when a **small** amount of a strong acid or base is added to it.



https://www.youtube.com/watch?v=LJmFbcaxDPE&feature=emb_logo

<https://prezi.com/h6obk63ws5ld/what-is-the-buffer-solution/>

<https://quizlet.com/77424730/buffer-solutions-flash-cards/>