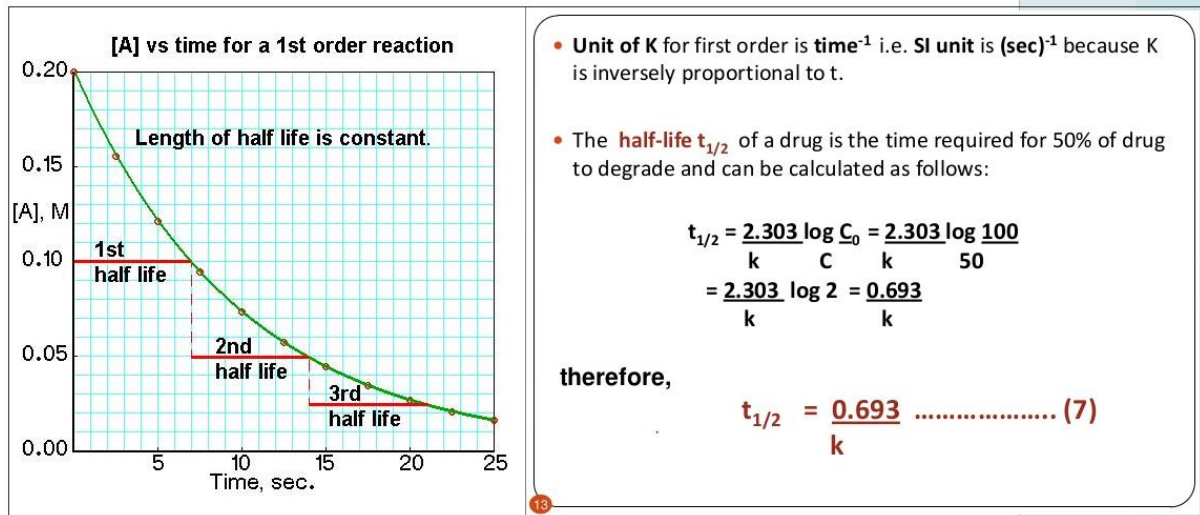


Half-life of first order reaction

Introduction: The half-life of reaction is the time it takes for half of the concentration of a reactant to be used up.



https://www.youtube.com/watch?v=e4qci9b2d3U&feature=emb_logo

<https://prezi.com/vb4mrwmur1nd/144-change-of-concentration-with-time/>

<https://quizlet.com/36365232/rates-of-chemical-reactions-flash-cards/>