

Coordinate (Dative Covalent) Bonding

https://www.youtube.com/watch?v=zLCU9Je7gQ4&feature=emb_logo

Introduction:

In normal covalent bonding, each atom involved contribute an equal number of electron(s) to the bond. However, in coordinate or dative covalent bonding, one atom of a species contributes the pair of electrons to the bond.

<https://www.slideshare.net/sqjafery/co-ordinate-covalent-bond?ref=https://notesmaster.com/en/group/caribbean/1556-cape-covid19-support/24480-coordinate-dative-covalent-bonding>